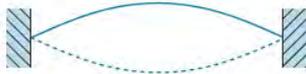


## G482: Electrons, waves and photons

Define	
<b>define</b> the <i>coulomb</i>	The SI unit of electrical charge. One Coulomb is defined as the amount of charge that passes in 1 second when the current is 1 ampere.
<b>define</b> <i>potential difference</i> (p.d.);	Energy transfer per unit charge from electrical to other forms.
<b>define</b> the <i>volt</i>	1 Volt is equal to 1 Joule per Coulomb ( $JC^{-1}$ ) The potential difference across a component is 1 volt when you convert 1 joule of energy moving 1 coulomb of charge through the component.
<b>define</b> <i>electromotive force</i> (e.m.f.) of a source such as a cell or a power supply	Energy transfer per unit charge from chemical/other to electrical form. Measured in V or $JC^{-1}$
<b>define</b> <i>resistance</i>	Resistance = Potential difference/current . Accept voltage instead of p.d.; ratio of voltage to current; voltage <u>per</u> (unit) current ( $VA^{-1}$ )  resistance = p.d./current
<b>define</b> the <i>ohm</i>	A component has a resistance of 1 ohm if a potential difference of 1 volt makes a current of 1 amp flow through it.
<b>define</b> <i>resistivity</i> of a material	Resistivity is equal to the product of the resistance and cross sectional area divided by the length. $P=RA/l$
<b>define</b> the kilowatt-hour (kW h) as a unit of energy	A unit of energy equal to 36 MJ or 1kW for 1h (a unit of) energy equal to 3.6 MJ or 1 kW for 1 h/AW
<b>define</b> and <b>use</b> the terms <i>displacement, amplitude, wavelength, period, phase difference, frequency and speed of a wave</i>	Displacement-how far a point on the wave has moved from its undisturbed position. Displacement-Distance from the mean position expressed as a vector Amplitude- Maximum displacement Wavelength-Distance between neighbouring identical points Period-Time taking for one complete oscillation of a particle Phase Difference- The fraction of a cycle between the oscillations of two particles Frequency-Number of waves passing a point per unit time Speed-Distance travelled by the wave per unit time
<b>define</b> the terms <i>nodes</i> and <i>antinodes</i>	Node-When the amplitude is always zero Antinode-When the amplitude is always at its maximum possible value
<b>define</b> and <b>use</b> the terms <i>fundamental mode of vibration</i> and <i>harmonics</i>	Simplest pattern of movement and has the lowest possible frequency band and the longest wavelength Harmonics are different modes of vibration of a wave with increasing frequency and decreasing wavelength

## G482: Electrons, waves and photons

	mode	wavelength	frequency
	first	$2L$	$\frac{v}{2L}$
	second	$L$	$\frac{v}{L}$
	third	$\frac{2L}{3}$	$\frac{3v}{2L}$
	fourth	$\frac{L}{2}$	$\frac{2v}{L}$
<b>define</b> and <b>use</b> the electronvolt (eV) as a unit of energy	<p>Electronvolt is defined as the kinetic energy gained by an electron when it is accelerated through a potential difference of 1 volt.</p> <p>1 eV is gained or lost when an electron moves through a potential difference of 1V</p> <p>Energy acquired by an electron accelerated through a p.d of 1V. <math>1eV = 1.6 \times 10^{-19}</math></p> <p><b>an eV is the energy to accelerate/move an electron through a p.d. of 1</b></p> <p><math>1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}</math> – NOT a definition</p>		
<b>define</b> and <b>use</b> the terms <i>work function</i> and <i>threshold frequency</i>	<p><b>Work function-</b> the <u>minimum</u> energy required to release an electron from the <u>surface</u> of a material</p> <p><b>Threshold frequency-</b> the lowest possible frequency of a photon that will cause an electron to be emitted from the material.</p>		
<b>Define the term intensity</b>	<p>intensity is the (incident) energy <u>per</u> unit area <u>per</u> second</p> <p>Intensity is the rate of flow of energy per unit area at right angles to the direction of travel of the wave. It is measured in <math>\text{Wm}^{-2}</math>.</p> <p>-number of photoelectrons emitted per second is proportional to the intensity of the radiation.</p>		

## G482: Electrons, waves and photons

State	
<b>state</b> what is meant by the term <i>mean drift velocity</i> of charge carriers	The average distance travelled by the charge carriers along the wire per second
<b>state</b> and <b>use</b> Ohm's law	Provided the temperature is constant, the current through an ohmic conductor is directly proportional to the potential difference across it. $R=V/I$ -This means the resistance is constant.
<b>state</b> Kirchhoff's second law and appreciate that this is a consequence of conservation of energy	Energy is conserved (sum) of e.m.f's = sum/total of voltages/p.d.s in/around a (closed) loop (in a circuit)
<b>state</b> typical values for the wavelengths of the different regions of the electromagnetic spectrum from radio waves to $\gamma$ -rays	Visible 600-400nm ( $5 \times 10^{-7}$ ) UV-A 400-315nm UV-B 315-260nm UV-C 260-100nm Radio Waves ( $10^{-1}$ to $10^6$ ) Micro waves ( $10^{-3}$ to $10^{-1}$ ) Infrared ( $7 \times 10^{-7}$ to $10^{-3}$ ) x-rays ( $10^{-13}$ to $10^{-8}$ ) Gamma rays ( $10^{-16}$ to $10^{-10}$ )
<b>state</b> that all electromagnetic waves travel at the same speed in a vacuum (this is one property of electromagnetic waves not shared by other waves)	
<b>state</b> that light is partially polarised on reflection	If you direct a beam of unpolarised light at a reflective surface then view the reflected ray through a polarising filter, the intensity of the light leaving the filter changes with the orientation of the filter. Intensity changes because light is partially polarised when it is reflected.
<b>state</b> and <b>use</b> the principle of superposition of waves	When two or more waves meet (at a point and interfere), The (resultant) <u>displacement</u> equals the (vector) <u>sum</u> of the <u>displacements</u> of each wave.
<b>state</b> what is meant by constructive interference and destructive interference	Interference is when (two) waves meet/combine/interact/superpose (at a point) and there is a change in overall intensity/displacement. A crest plus a crest makes an even bigger crest, this is the same with two troughs, these two are both constructive interference. However when a crest hits a trough of equal size it gives nothing, this is destructive interference.
<b>state</b> that a photon is a quantum of energy of electromagnetic radiation. Einstein suggested that photons are wave packets of electromagnetic radiation that carried the energy. He said that photons act as particles can either transfer all or none of its energy when colliding with another particle. Max Planck said that EM waves can only be released in packets called quanta. So a photon is a single quantum of EM radiation.	
<b>state</b> that energy is conserved when a photon interacts with an electron. Photons either	

## G482: Electrons, waves and photons

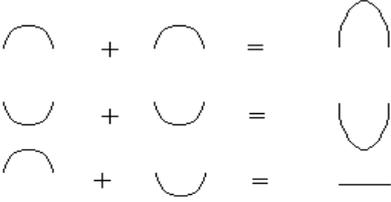
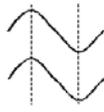
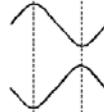
transfer all or none of their energy when they interact with another particle.
state that the charge carriers in an electrolyte are ions. Once molten the liquid conducts, the positive and negative ions are the charged carriers. Same thing in ionic solution.
state that the charge carriers moving through wires are electrons. Wires are made of metal, and in metal the charged carriers are free electrons-they're the ones in the outer shell of each atom.

Select and use (Refer to the Formula sheet)	
select and use the equation	$\Delta Q = I\Delta t$
select and use the equation	$I = Anev$
select and use the equation	$W = VQ$
select and use the equation for resistance	$R = V/I$
select and use the equation	$R = \rho L/A$
select and use power equations	$P = VI$ $P = I^2R$ and $P = V^2/R$
select and use the equation	$W = IVt$
select and use the equation for the total resistance of two or more resistors in series	
select and use the equation for the total resistance of two or more resistors in parallel	
select and use the equations and	e.m.f. = $I(R + r)$ , e.m.f. = $V + Ir$
select and use the potential divider equation:	$V_{out} = \frac{R_2 \times V_{in}}{R_1 + R_2}$
select and use the wave equation	$v = f\lambda$
select and use the equation for electromagnetic waves	$\lambda = ax/D$
select and use the equation	$d\sin\theta = n\lambda$
select and use the equations for the energy of a photon:	$E = hf$ and $E = hc/\lambda$
select, explain and use Einstein's photoelectric equation	$hf = \phi + KE_{max}$
select and apply the de Broglie equation	$\lambda = h/mv$

Recall and use	
recall and use the elementary charge $e = 1.6 \times 10^{-19}C$	
recall and use appropriate circuit symbols as set out in SI Units, Signs, Symbols and Abbreviations (ASE, 1981) and Signs, Symbols and Systematics (ASE, 1995) interpret and draw circuit diagrams using these symbols	
recall and apply Malus's law for transmitted intensity of light from a polarising filter.	The intensity of light transmitted through a polarising filter is equal to $I_0 \cos^2 \theta$

Use and apply	
use the relationships: intensity = power/cross-sectional area and intensity $\propto$ amplitude <sup>2</sup>	
use the equation: separation between adjacent nodes (or antinodes) = $\lambda/2$	
use the transfer equation: $eV = 1/2mv^2$ for electrons and other charged particles	
use the relationships	$hf = E_1 - E_2$ and $hc = E_1 - E_2 / \lambda$
apply Kirchhoff's first and second laws to circuits	

## G482: Electrons, waves and photons

	<p>constructive interference reinforcement</p> <p>constructive interference reinforcement</p> <p>destructive interference cancellation</p>	<p>a) Waves in phase</p>  <p>b) Waves out of phase</p> 	<p>Result if waves superpose</p>  <p>Result if waves superpose</p> 
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**apply** graphical methods to illustrate the principle of superposition

### Describe

**describe** how an ammeter may be used to measure the current in a circuit

Must be put in series to measure the current. You can measure the current in a circuit using an ammeter which must be attached in series so that the current running through the component is the same going through the ammeter.

**describe** Kirchoff's first law and appreciate that this is a consequence of conservation of charge

sum of total current into a junction equals the sum of total current out  
conservation of charge

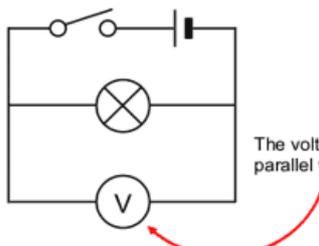
**describe** the difference between conductors, semiconductors and insulators in terms of the number density  $n$

-In conductors, such as metal, the charge carriers are free electrons. So there are loads of charge carries making  $n$  big. Therefore the drift velocity can be small for a high current.

-In semiconductors there are fewer charge carries so  $n$  is smaller. This means that the drift velocity has to be higher to get the same current as conductors.

-A perfect insulator wouldn't have any charge carries so  $n$  would be 0. Real insulators have very small numbers of charged carried so very small  $n$  figures.

**describe** how a voltmeter may be used to **determine** the p.d. across a component



The voltmeter must be put in parallel

The voltmeter is in parallel with the lamp

**describe** the difference between e.m.f. and p.d. in terms of energy transfer

Potential difference is transferring electrical energy into other forms (heat, light, sound). EMF is transferring other forms of energy into electrical (chemical energy stored in a battery)

EMF is the amount of electrical energy the battery produces for each coulomb of charge. Measured in volts.

-Whereas the p.d. is the energy transferred when one coulomb of charge flows through a load resistance.

**describe** the  $I-V$  characteristics of a resistor at constant temperature, filament lamp and light-emitting diode (LED)

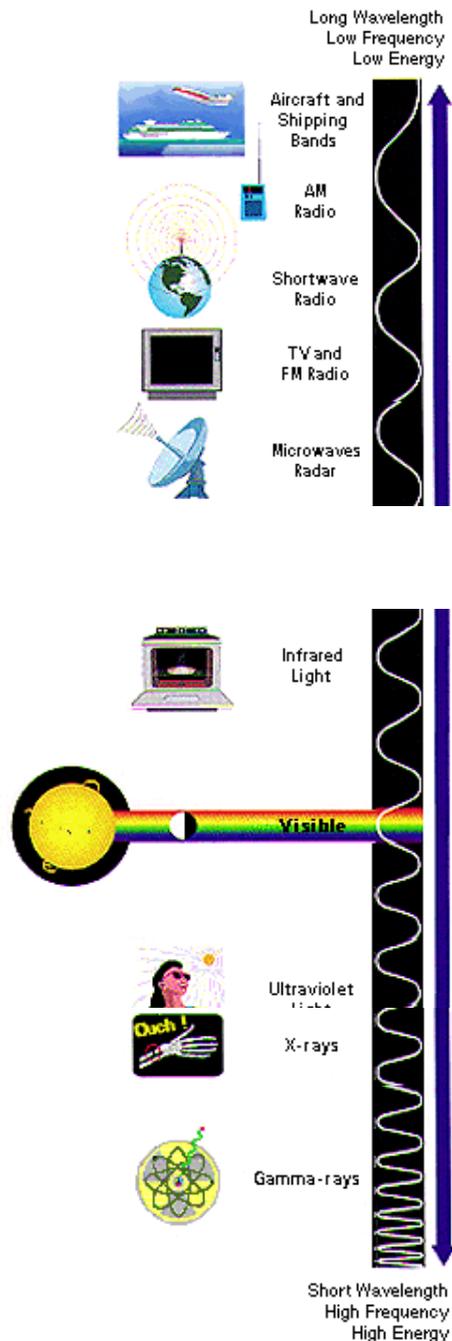
Resistor follows ohms law so straight line through origin. Filament Lamp is non-ohmic as its temperature varies so the line should go up and then curve so the gradient = 0. LED is non-ohmic as well, so no current should flow until a certain threshold voltage, so flat line to indicate no current, then a upwards line to show current is now flowing.

## G482: Electrons, waves and photons

	<p style="text-align: center;">Ohmic conductor      Semiconductor diode      Filament lamp</p>
<p><b>describe</b> an experiment to obtain the <math>I-V</math> characteristics of a resistor at constant temperature, filament lamp and light-emitting diode (LED)</p>	<p>Ammeter, Resistor/Filament Lamp/LED and Potentiometer in series. Place voltmeter in parallel with the component being tested. Limit the current flowing by varying the potentiometer accordingly, taking current and potential difference readings respectively</p>
<p><b>describe</b> the uses and benefits of using light emitting diodes (LEDs).</p>	<p><b>Advantages of LED over a filament lamp in a torch-Draws lower current, light lasts longer, LEDs more efficient at converting electrical energy into light, more robust and a longer working life.</b></p>
<p><b>describe</b> how the resistivities of metals and semiconductors are affected by temperature</p>	<p>Near room temperature, the resistivity of metals typically increases as temperature is increased, while the resistivity of semiconductors typically decreases as temperature is increased.</p>
<p><b>describe</b> how the resistance of a pure metal wire and of a negative temperature coefficient (NTC) thermistor is affected by temperature</p>	<p>Thermistors are <b>temperature sensitive resistors</b>. However, unlike most other resistive devices, the resistance of a thermistor decreases with increasing temperature. In a pure metal a greater resistance slows the flow of electrons so a smaller current flows as temperature increases.</p>
<p><b>describe</b> power as the rate of energy transfer</p>	<p>power is the rate at which energy is transferred, used, or transformed <math>\text{J s}^{-1}</math></p>
<p><b>describe</b> how the resistance of a light dependent resistor (LDR) depends on the intensity of light</p>	<p><b>Resistance decreases with increase in light intensity</b>  <b>LDR must be shielded or be at some distance from the lamp when it switches on as the light shining will cause it to switch the illumination off causing an on/off oscillation</b></p>
<p><b>describe</b> and explain the use of thermistors and light-dependent resistors in potential divider circuits</p>	<p>Thermistor/LDR can be used to provide an output voltage, which depends on temperature/light intensity.</p>
<p><b>describe</b> the advantages of using dataloggers to monitor physical changes</p>	<p><b>Continuous record for a very long time scale of observations</b>  <b>Can record very short timescale signals (at intervals)</b>  <b>Automatic recording/remote sensing</b>  <b>Data can be fed directly to a computer (for analysis)</b></p>
<p><b>describe</b> and distinguish between progressive longitudinal and transverse waves</p>	<p><b>Longitudinal = oscillations/vibration of <u>particles/medium</u> in direction of travel of the wave e.g. sound</b>  <b>Transverse = oscillations/vibration of <u>particles/medium</u> (in the plane) at right angles to the direction of travel of the wave e.g. surface water, string, electromagnetic</b></p>

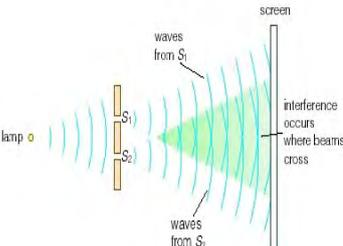
## G482: Electrons, waves and photons

describe differences and similarities between different regions of the electromagnetic spectrum

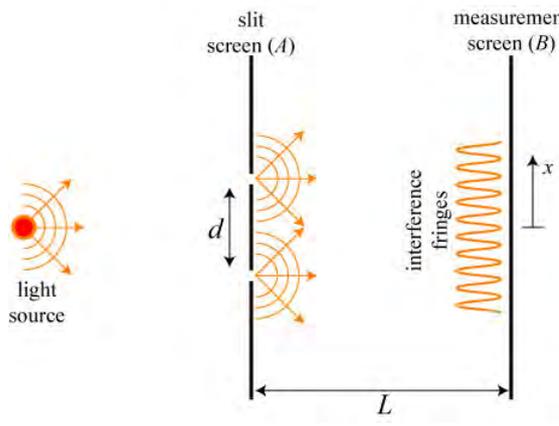
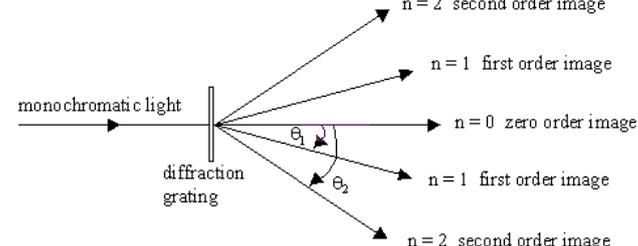


Electromagnetic radiation can be described in terms of a stream of photons, which are massless particles each travelling in a wave-like pattern and moving at the speed of light. Each photon contains a certain amount (or bundle) of energy, and all electromagnetic radiation consists of these photons. The only difference between the various types of electromagnetic radiation is the amount of energy found in the photons. Radio waves have photons with low energies, and gamma-rays have photons with the greatest energies.

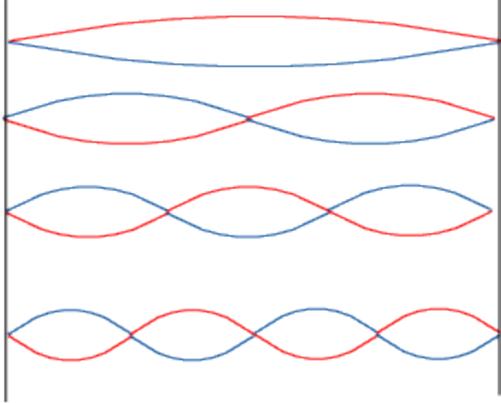
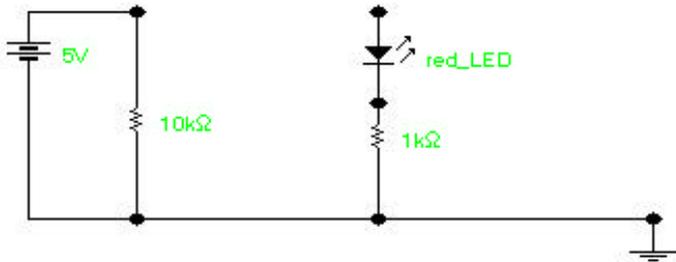
## G482: Electrons, waves and photons

<p><b>describe</b> some of the practical uses of electromagnetic waves</p>	<p>Radio stations. Radio waves are emitted by stars and gases in space.          Microwaves in space are used by astronomers to learn about the structure of nearby galaxies.          Our skin emits infrared light and In space, IR light maps the dust between stars.          Visible radiation is emitted by everything from fireflies to light bulbs to stars ... also by fast-moving particles hitting other particles. It's the light the human eye can see.          The Sun, Stars and other "hot" objects in space emit UV radiation.          Hot gases in the Universe also emit X-rays. They are used in scanning the bones in the body.          Radioactive materials (some natural and others made by man in things like nuclear power plants) can emit gamma-rays. The biggest gamma-ray generator of all is the Universe! It makes gamma radiation in all kinds of ways.</p>
<p><b>describe</b> the characteristics and dangers of UV-A, UV-B and UV-C radiations and <b>explain</b> the role of sunscreen</p>	<p>filters out/blocks/reflects/absorbs UV(-B)          UV-A causes tanning or skin ageing ; most of (99%) uv light;          400-315 nm          UV-B causes damage or sunburn or skin cancer; 315-260 nm          UV-C is filtered out by atmosphere/ozone layer; 260-100 nm</p>
<p><b>describe</b> experiments that demonstrate two source interference using sound, light and microwaves</p>	<p>Two speakers playing the same note, hear constructive (loud) and destructive (quiet) interference.          Light or microwaves pointed towards two slits constructive and destructive interference can be observed.</p>
<p><b>describe</b> constructive interference and destructive interference in terms of path difference and phase difference</p>	<p>If the path difference between two light waves is <math>(m+1/2)\lambda</math>, then the interference between them will be destructive.          For constructive interference, path difference between two waves is <math>m\lambda</math></p>
<p><b>describe</b> the Young double-slit experiment and <b>explain</b> how it is a classical confirmation of the wave-nature of light</p>	<div style="display: flex; align-items: center;"> <div style="flex: 1;">  </div> <div style="flex: 1; padding-left: 20px;"> <p>Monochromatic source sent through 2 slits which diffracts the source, since they diffract it shows wave nature.</p> </div> </div>

## G482: Electrons, waves and photons

<p>describe an experiment to determine the wavelength of monochromatic light using a laser and a double slit</p>	<p><math>\lambda = \frac{ax}{D}</math> <math>a=d</math>=slit separation <math>x</math>=fringe separation  <math>D=L</math>=distance between slit and screen</p> 
<p>describe the use of a diffraction grating to determine the wavelength of light (the structure and use of a spectrometer are not required)</p>	 <p><math>d \sin \theta = n\lambda</math> <math>n</math>=order of diffraction</p>
<p>describe the similarities and differences between progressive and stationary waves</p>	<p><b>Progressive-A wave which transfers energy as a result of oscillations(of the source/medium/particles through which the energy is travelling)</b>  <b>A progressive wave transfers shape/information from one place to another.</b>  <b>Every point on a progressive wave has the same amplitude.</b>  <b>Every point on a progressive wave oscillates.</b>  <b>All points on a progressive wave (in 1 wavelength) have different phase</b></p> <p><b>Stationary-A wave which stores energy (traps energy in pockets)</b>  <b>The shape does not move along – does not transfer shape/information</b>  <b>Stationary wave has nodes and antinodes.</b>  <b>Stationary wave = the incident wave is reflected at the end of the pipe. Reflected wave interferes/superposes with the incident wave to produce (a resultant wave with) nodes and/or antinodes</b></p>

## G482: Electrons, waves and photons

<p><b>describe</b> experiments to demonstrate stationary waves using microwaves, stretched strings and air columns</p>	<div style="display: flex; align-items: center;"> <div style="flex: 1;"> <p>Fundamental 1st Harmonic</p> <p>First Overtone 2nd Harmonic</p> <p>Second Overtone 3rd Harmonic</p> <p>Third Overtone 4th Harmonic</p> <p>And so on...</p> <p style="text-align: center;">⋮</p> </div> <div style="flex: 2;">  </div> </div> <p>Open/closed end air column or string vibrated with a fixed point</p>
<p><b>describe</b> the particulate nature (photon model) of electromagnetic radiation</p>	<p>A photon is a quantum/lump/unit/packet/particle of (e-m) energy/light</p>
<p><b>describe</b> an experiment using LEDs to estimate the Planck constant <math>h</math> using the equation</p>	
<p><b>describe</b> and <b>explain</b> the phenomenon of the photoelectric effect</p>	<p>A photon is absorbed by an electron in a metal surface causing an electron to be emitted. Energy is conserved. Only photons with energy above the work function energy will be emitted. Energy = work function + Max KE of electron. The work function is the minimum energy required to release an electron from the surface.</p> <p>A clean zinc plate is mounted on the cap of a gold leaf electroscope where the plate is initially charged negatively. Shine a UV light on the plate and watch the gold leaf collapse as charge leaves the plate, indicating the emission of electrons.</p> <p>Work function energy is the <u>minimum</u> energy to release an electron from the surface.</p> <p>Number of electrons emitted also depends on light intensity.</p> <p>Emission is instantaneous.</p> <p>energy of the infra-red photon is less than the work function of the metal surface</p> <p>State: emission of electron(s) from a metal (surface) when photon(s)/light/uv/em radiation are incident (on surface)</p>
<p><b>describe</b> the origin of emission and absorption line spectra</p>	<p>(Emission) Line spectrum = light emitted from (excited) atoms produces a line spectrum a series of</p>

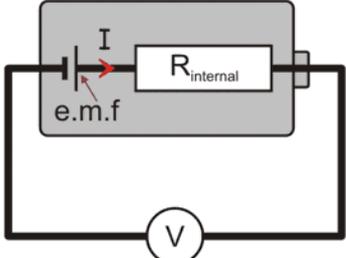
## G482: Electrons, waves and photons

	<p>(sharp/bright/coloured) lines against a dark background. Absorption spectrum is a series of <u>dark</u> lines (appears against a bright background/within a continuous spectrum)</p> <p style="text-align: center;">Hydrogen Absorption Spectrum</p>  <p style="text-align: center;">Hydrogen Emission Spectrum</p>  <p style="text-align: center;">400nm <span style="margin-left: 300px;">700nm</span></p> <p style="text-align: center;">H Alpha Line 656nm Transition N=3 to N=2</p>
The difference between the directions of conventional current and electron flow	<p>current moves from + to – (of battery in circuit) and electrons move from – to +</p>

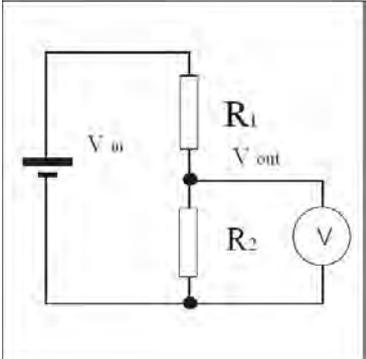
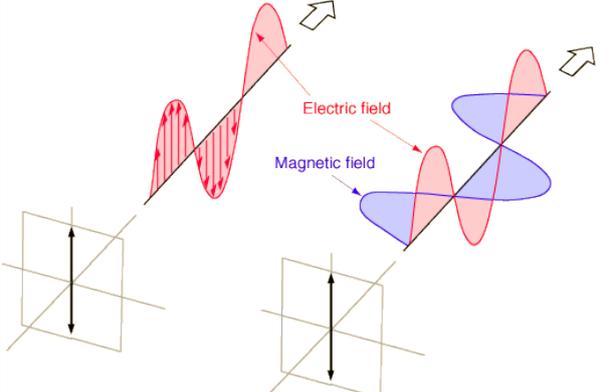
<b>Determine</b>	
<b>determine</b> the correct fuse for an electrical device	<p>The next number of amps above the current of the circuit. (smallest value above possible) You can usually only get fuses with ratings of 3amps, 5amps or 13amps.</p>
<b>determine</b> the standing wave patterns for stretched string and air columns in closed and open pipes	
<b>determine</b> the speed of sound in air from measurements on stationary waves in a pipe closed at one end	<p>Sound waves in the tube are in the form of <u>standing waves</u>, and the <u>amplitude</u> of vibrations of air is zero at equally spaced intervals along the tube.. The powder is caught up in the moving air and settles in little piles at these nodes. The distance between the piles is one half <u>wavelength</u> <math>\lambda/2</math> of the sound. By measuring the distance between the piles, the wavelength <math>\lambda</math> of the sound in air can be found. If the frequency <math>f</math> of the sound is known, multiplying it by the wavelength gives the speed of sound <math>c</math> in air: <input style="width: 50px; height: 15px;" type="text"/></p>

<b>Explain</b>	
explain that electric current is a net flow of charged particles	<p>There is a current when charged particles flow past a point in a circuit. Current is the rate of flow of charge. Current in a wire is like water flowing in a pipe. The amount of water that flows depends on the flow rate and the time. Current is the rate of flow of charge.</p>
explain that electric current in a metal is due	<p>Wires are made from metal. The metal</p>

## G482: Electrons, waves and photons

<p>to the movement of electrons, whereas in an electrolyte the current is due to the movement of ions</p>	<p>contains a sea of delocalised electrons which move in random directions. When a cell is connected to the wire and electrical force is applied to the electrons making them 'drift'. They still move in random directions however they have an overall velocity or movement, creating a current. This can happen with ions in an electrolyte too.</p>
<p>explain what is meant by conventional current and electron flow</p>	<p>The direction of the current is from the positive terminal to the negative. However electrons are what flow in metals and are negatively charged and therefore flow from negative to positive</p>
<p>explain how a fuse works as a safety device</p>	<p>The fuse needs to have a current rating big enough to cover the initial current in the circuit. If the current gets too big, it melts the wire which breaks the circuit.</p>
<p>explain that all sources of e.m.f. have an internal resistance</p>	<p>All sources of emf have what is known as INTERNAL RESISTANCE (<math>r</math>) to the flow of electric current. The internal resistance of a fresh battery is usually small but increases with use. Thus the voltage across the terminals of a battery is less than the emf of the battery. <math>E = I(R + r)</math></p>  <p>The diagram shows a rectangular circuit loop. On the left vertical wire, there is a battery symbol with 'e.m.f.' written below it. A red arrow labeled 'I' points from the positive terminal to the negative terminal. On the top horizontal wire, there is a box labeled 'R<sub>internal</sub>'. On the bottom horizontal wire, there is a circle with a 'V' inside, representing a voltmeter.</p>

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<p>explain how a potential divider circuit can be used to produce a variable p.d</p>	<p>If you use two fixed resistors in series and connect the ends to a voltage supply then the voltage between both ends and the mid point where they connect will be in proportion their resistor value. It is possible to vary the midpoint voltage by changing one of the resistors. This can be done readily with a variable resistor which than then be altered to get any value between 0-MaxV By adjusting the value of R1, the potential dropped</p> 
<p>explain what is meant by reflection, refraction and diffraction of waves such as sound and light</p>	<p>Reflection-Bouncing back of wave from a surface Refraction-Change in direction of a wave as it crosses and interface between two materials where its speed changes Diffraction-Spreading of a wave when it passes through a gap or past the edge of an object</p>
<p>explain the meaning of the term <i>terminal p.d.</i>;</p>	<p>The terminal p.d. of a source is the potential difference across its terminals</p>
<p>explain what is meant by plane polarised waves and understand the polarisation of electromagnetic waves</p>	 <p>Transverse waves/vibrations in plane are normal to the direction of energy propagation. Oscillations are in one direction only</p>
<p>explain that polarisation is a phenomenon associated with transverse waves only</p>	<p>Electromagnetic waves are transverse waves so can be polarised whereas sound waves cannot since they are not transverse.</p>
<p>explain the terms <i>interference</i>, <i>coherence</i>, <i>path difference</i> and <i>phase difference</i></p>	<p>Interference-When two waves meet at a point Coherence-Constant phase difference between the two waves Path difference-of any point in an</p>

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	<p>interference pattern of waves is the difference between the distance travelled by each wave from their source to that point</p> <p><b>Phase difference</b>- difference in velocity of similar points in two waves expressed as an angle Coherence = <u>constant</u> phase relationship or are <u>continuous</u> and have the same <math>f/\text{period}/\lambda</math></p>
<p>explain the advantages of using multiple slits in an experiment to find the wavelength of light.</p>	
<p>explain that the photoelectric effect provides evidence for a particulate nature of electromagnetic radiation while phenomena such as interference and diffraction provide evidence for a wave nature</p>	<p>There is a threshold frequency which suggests particle nature, as the wave theory states that photoelectric emission should happen as long as the light is bright enough. However this is not the case. Diffraction and interference are wave properties, suggesting that electromagnetic radiation has wave nature.</p>
<p>explain the formation of stationary (standing) waves using graphical methods</p>	<p>(open end tube and speaker) Using a tube with one end closed and a loud speaker, the incident wave is reflected at the end of the pipe and it interferes with the incident wave to produce a resultant wave</p> <p>(string and oscillator) The incident wave is reflected at the fixed end of the wire, the reflected wave interferes with the incident wave to produce a resultant wave with nodes and antinodes</p>
<p>explain and use Einstein's photoelectric equation <math>hf = \phi + K_{E_{max}}</math></p>	<p>Individual photons are absorbed by individual electrons in the metal's surface. These electrons must absorb sufficient energy to overcome the work function energy of the metal. The number of electrons emitted depends on light intensity as emission is instantaneous.</p> <p>Infra-red does not have enough energy to cause photoelectric emission it is less than the work function</p>
<p>explain why the maximum kinetic energy of the electrons is independent of intensity and why the photoelectric current in a photocell circuit is proportional to intensity of the incident radiation</p>	<p><math>hf = \phi + K_{E_{MAX}}</math> Therefore independent of intensity.</p> <p>The larger the intensity, the greater the number of photons emitted, therefore releasing more electrons generating a larger current.</p>
<p>explain electron diffraction as evidence for the wave nature of particles like electrons</p>	<p>Electron diffraction refers to the wave nature of electrons by firing electrons at a sample and observing the resulting interference pattern. This phenomenon is commonly known as the wave-particle duality, which states that the behaviour of a particle of</p>

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	<p>matter (in this case the incident electron) can be described by a wave.</p> <p>Diffraction is a property unique to waves. If electrons can be diffracted then they are behaving as waves</p>
explain that electrons travelling through polycrystalline graphite will be diffracted by the atoms and the spacing between the atoms	Graphite, because of its layered structure, can act as a diffraction grating with very small slit diameter.
explain that the diffraction of electrons by matter can be used to determine the arrangement of atoms and the size of nuclei.	Slow moving electrons, electrons with wavelengths ( $E=hf$ ) of the order of magnitude of the structure (or nuclei) can be used to probe the properties of atomic structures.
explain how spectral lines are evidence for the existence of discrete energy levels in isolated atoms, ie in a gas discharge lamp	Photon produced by electron moving between levels Photon energy equal to energy difference between levels
Explain how sunscreen protects the human skin	Filters out/blocks/reflects/absorbs UV (-B)
Explain why electrons can be emitted from a clean metal surface illuminated with bright UV light but never when IR is used, however intense	<u>Energy</u> of the infra-red photon is less than the <u>work function</u> of the metal surface
Explain what is meant by the de Broglie wavelength of an electron	Electrons are observed to behave as waves/show wavelike properties where the electron wavelength depends on its speed/momentum
Explain what is meant by a continuous spectrum	<u>All</u> wavelengths/frequencies are present (in the radiation)

Maths	
calculate energy in kW h and the cost of this energy when solving problems	
solve circuit problems involving series and parallel circuits with one or more sources of e.m.f	
derive from the definitions of speed, frequency and wavelength, the wave equation $v = f\lambda$	$v = \frac{x}{t}$ and $f = \frac{1}{t}$ then $v = \frac{x}{\frac{1}{f}}$ Wavelength $\lambda$ is the displacement $x$ between subsequent wave peaks therefore $v = f\lambda$
draw a simple potential divider circuit	